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Aldehydes can be oxidized to carboxylic acids; ketones cannot. The Tollens test and Benedicts test take advantage of this. Both tests involve reagents which react with aldehydes to produce a visible product. No reaction occurs with ketones. 18 See the solution in the text book (page A-30).

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sour and cause certain
dyes to change color.
• Bases taste bitter and
feel soapy. • Arrhenius
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bases: • An acid is a
substance that, when
dissolved in water,
increases the
concentration of H^+
ions.

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Name: Answer Key Due
Date: 3/1/2018 1. (5
pts) Circle all the
aromatic molecules.
Put an "X" through the
antiaromatic
molecules. Leave alone
all nonaromatic
molecules. (1 or 2
incorrect, ok, but after
that start taking partial
points off)

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The study of the
transfers of energy as
heat that accompany
ch.... The energy
absorbed or released
as heat in a chemical

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or physi.... A measure of the average kinetic energy of the particles in a....

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...

The probability for all the particles to be on one side is $\frac{1}{32}$. $\frac{1}{32}$. This probability is noticeably lower than the $\frac{1}{8}$ $\frac{1}{8}$ result for the four-particle system. The conclusion we can make is that the probability for all

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the particles to stay in only one part of the system will decrease rapidly as the number of particles increases, and, for instance, the probability for all molecules of gas to ...

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be used for the treatment of diseases and for reducing the suffering from pain are called medicines or drugs. The branch of science which makes use of chemicals for the treatment of diseases [therapeutic effect] is called chemotherapy.

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Chapter

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to the Problems IUPAC
names for each

compound. Specify the
configuration in (c). O

O C₆H₅\CHO(a)

7<\CHO (b) (C) 5CH₃

(R)-2-Phenylpropanal1

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fi-Cyclohexanedione
Problem 16.1 Write the 3,3
-Dimethylbutanal
Problem 16.2 Write structural
formulas for all
aldehydes with the
molecular formula
 $C_6H_{12}O$ and give each
its IUPAC name.

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which the bonding
electrons are shared
equally by the bonded
atoms, resulting in a
balanced distribution of

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If the pH of tomato juice is 4.0, what is the hydroxide ion concentration in this solution? 2. If the pH of an acid rain storm is approximately 3.0, how many times greater is the (H') in the rain than in a cup of coffee having a pH of 5.0? 3.

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